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CONFIRMATION NO. APPLICATION NO. FILING DATE FIRST NAMED INVENTOR ATTORNEY DOCKET NO. 10/034,094 12/27/2001 Sekharipuram R. Narayanan 06816-044003 2464 **EXAMINER** 20985 7590 12/01/2005 FISH & RICHARDSON, PC TSANG FOSTER, SUSY N P.O. BOX 1022 ART UNIT PAPER NUMBER MINNEAPOLIS, MN 55440-1022 1745

DATE MAILED: 12/01/2005

Please find below and/or attached an Office communication concerning this application or proceeding.

	Applicatio	n No.	Applicant(s)	
	10/034,09	4	NARAYANAN ET AL.	
Office Action Summary	Examiner		Art Unit	
	Susy N. Ts	ang-Foster	1745	
The MAILING DATE of this comm Period for Reply	unication appears on the	cover sheet with the c	orrespondence ad	ddress
A SHORTENED STATUTORY PERIOD WHICHEVER IS LONGER, FROM THE - Extensions of time may be available under the provis after SIX (6) MONTHS from the mailing date of this c - If NO period for reply is specified above, the maximum - Failure to reply within the set or extended period for r Any reply received by the Office later than three monie earned patent term adjustment. See 37 CFR 1.704(b)	E MAILING DATE OF TH ions of 37 CFR 1.136(a). In no ever ommunication. In statutory period will apply and will eply will, by statute, cause the applichs after the mailing date of this con	IS COMMUNICATION nt, however, may a reply be tim l expire SIX (6) MONTHS from cation to become ABANDONE	N. nely filed the mailing date of this o D (35 U.S.C. § 133).	·
Status				
 Responsive to communication(s) This action is FINAL. Since this application is in conditional closed in accordance with the present the communication of the co	2b)⊠ This action is no on for allowance except	on-final. for formal matters, pro		e merits is
Disposition of Claims				•
4) Claim(s) 1-26 is/are pending in the 4a) Of the above claim(s) 7-25 is/ 5) Claim(s) is/are allowed. 6) Claim(s) 1-6 and 26 is/are rejected to 7) Claim(s) is/are objected to 8) Claim(s) are subject to research	are withdrawn from consi ed.			
9)☐ The specification is objected to by	the Evaminer			
10) The drawing(s) filed on is/a Applicant may not request that any o Replacement drawing sheet(s) include 11) The oath or declaration is objecte	re: a) accepted or b) □ bjection to the drawing(s) be ling the correction is require	e held in abeyance. See ed if the drawing(s) is obj	e 37 CFR 1.85(a). jected to. See 37 C	
Priority under 35 U.S.C. § 119				
 12) Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f). a) All b) Some * c) None of: 1. Certified copies of the priority documents have been received. 2. Certified copies of the priority documents have been received in Application No. 3. Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)). * See the attached detailed Office action for a list of the certified copies not received. 				
Attachment(s) 1) Notice of References Cited (PTO-892) 2) Notice of Draftsperson's Patent Drawing Review 3) Information Disclosure Statement(s) (PTO-1449 Paper No(s)/Mail Date	•	4) Interview Summary Paper No(s)/Mail Da 5) Notice of Informal P 6) Other:	ate	O-152)

DETAILED ACTION

Continued Examination Under 37 CFR 1.114

1. A request for continued examination under 37 CFR 1.114, including the fee set forth in 37 CFR 1.17(e), was filed in this application after final rejection. Since this application is eligible for continued examination under 37 CFR 1.114, and the fee set forth in 37 CFR 1.17(e) has been timely paid, the finality of the previous Office action has been withdrawn pursuant to 37 CFR 1.114. Applicant's submission filed on 9/12/2005 has been entered.

Response to Amendment

2. This Office Action is responsive to the amendment filed on 9/12/2005. Claims 1 and 26 have been amended. Claims 1-26 are pending. Claims 7-25 are withdrawn from further consideration as being directed to nonelected species. Claims 1-6, and 26 are rejected for reasons given below.

Specification

3. The amendment filed 9/12/2005 is objected to under 35 U.S.C. 132(a) because it introduces new matter into the disclosure. 35 U.S.C. 132(a) states that no amendment shall introduce new matter into the disclosure of the invention. The added material which is not supported by the original disclosure is as follows:

In paragraph 69 on page 18, starting at line 16, adding the phrase "into the catalyst material" in the first sentence constitutes new matter. Also in paragraph 69, adding the phrase "can also be used as the substance insoluble in the catalyst material" constitutes new matter.

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The original specification at paragraph 69 is limited to temporary introduction of a substance insoluble in the <u>catalyst</u>, not the <u>catalyst material</u> which comprises the <u>catalyst</u> and other components such as binder. In addition, the original specification at paragraph 69 states that the insoluble substance in the <u>catalyst</u> can be a surface active substance which will prevent particle agglomeration and can be volatilized at a higher temperature and does not disclose that the insoluble substance is in the <u>catalyst material</u>.

Applicant is required to cancel the new matter in the reply to this Office Action.

Claim Rejections - 35 USC § 112

4. The following is a quotation of the first paragraph of 35 U.S.C. 112:

The specification shall contain a written description of the invention, and of the manner and process of making and using it, in such full, clear, concise, and exact terms as to enable any person skilled in the art to which it pertains, or with which it is most nearly connected, to make and use the same and shall set forth the best mode contemplated by the inventor of carrying out his invention.

5. Claims 1-6 and 26 are rejected under 35 U.S.C. 112, first paragraph, as failing to comply with the written description requirement. The claim(s) contains subject matter which was not described in the specification in such a way as to reasonably convey to one skilled in the relevant art that the inventor(s), at the time the application was filed, had possession of the claimed invention.

With respect to instant claim 1 and its elected dependent claims, the original disclosure does not support the limitation of introducing a substance in the <u>catalyst material</u>, wherein the substance is insoluble in the <u>catalyst material</u> and subsequently removing the insoluble substance from the <u>catalyst material</u> to increase a surface area of the <u>catalyst</u> compared to a surface area of the <u>catalyst</u> in the <u>catalyst material</u> without introducing and removing the substance.

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Paragraph 69 at page 18 of the instant original specification states:

[0 0 6 9] (1) Temporary introduction of a substance insoluble in the catalyst and removable by subsequent leaching may be used to increase the surface area of the catalyst. Such a substance can be a surface active substance which will prevent particle agglomeration and can be volatilized at a higher temperature. Non-ionic surfactants may be preferable. Metals such as zinc, aluminum, or tin incorporated in the catalyst during preparation can also serve this function.

The use of the term "catalyst" and "catalyst material" appears to mean two different things in the present specification and in the claims. The Examiner is interpreting the term "catalyst material" to mean a catalyst mixture comprising at least a catalyst and a binder in light of claim 2 and applicant's specification. Nowhere in paragraph 69 in the original specification does it state that the insoluble substance is introduced into the <u>catalyst material</u>. Original paragraph 69 states that the insoluble substance is temporarily introduced in the <u>catalyst</u>.

With respect to claim 26, the limitations of "preparing a catalyst material to comprises a catalyst and at least a metal of zinc, aluminum, or tin which is insoluble; and subsequently removing the metal from the catalyst material by leaching to increase a surface area of the catalyst compared to a surface area of the catalyst without removing the metal" are not in the original disclosure for the same reasons given above.

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6. The following is a quotation of the second paragraph of 35 U.S.C. 112:

The specification shall conclude with one or more claims particularly pointing out and distinctly claiming the subject matter which the applicant regards as his invention.

7. Claims 1-6 and 26 are rejected under 35 U.S.C. 112, second paragraph, as being indefinite for failing to particularly point out and distinctly claim the subject matter which applicant regards as the invention.

With respect to claim 26, the limitations of "preparing a catalyst material to comprises a catalyst and at least a metal of zinc, aluminum, or tin which is insoluble; and subsequently removing the metal from the catalyst material by leaching to increase a surface area of the catalyst compared to a surface area of the catalyst without removing the metal" are indefinite because it is unclear how the surface area of the <u>catalyst</u> is increased by leaching the metal from the <u>catalyst material</u> since the <u>catalyst material</u> is a mixture of several components such as <u>catalyst</u>, perfluorovinylether sulfonic acid and polytetrafluoroethylene (see paragraph 14 of the specification and instant claim 2).

Applicant states on page 9 of the remarks that the mixing of zinc, aluminum, and tin in other materials and leaching them out are known in the art. It is unclear to the Examiner how mixing in zinc, aluminum, and tin into a <u>catalyst material</u> comprising a <u>catalyst</u>, a binder, and perfluorovinylether sulfonic acid and subsequently leaching of zinc, aluminum, and tin would create voids in the <u>catalyst</u> since mixing of the metals (zinc, tin, aluminum) into the <u>catalyst</u> material only creates a physical association of the metals with the <u>catalyst</u> and not a chemical one. Since the insoluble substances are not introduced into the <u>catalyst</u> in the <u>catalyst material</u> by mixing, they would not be able to create voids in the <u>catalyst</u> itself after the step of removing

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the insoluble substance contrary to applicant's assertions on page 10 of the remarks filed on 9/12/2005.

Claims 1-6 are also indefinite for the same reason given above for claim 26 because it is unclear how the surface area of a catalyst is increased by removing an insoluble substance from a catalyst material.

Response to Arguments

Applicant's arguments filed 9/12/2005 have been fully considered but they are not 8. persuasive.

Applicant appears to misunderstand that the terms "catalyst material" and "catalyst" are distinct as presently used in the specification and in the instant claims.

The specification discloses increasing the surface area of the catalyst, and does not disclose increasing the surface area of the catalyst material which comprises catalyst, binder, and other components used in a catalyst layer of the fuel cell electrode.

Applicant asserts in the remarks filed on 9/12/2005 that the word "catalyst" in line 22 on page 18 should be "catalyst material" when reading the paragraph as a whole. The Examiner disagrees since as evidenced by instant claim 4, the <u>catalyst</u> comprises platinum and ruthenium. The <u>catalyst</u> is a metal or metal alloy, whereas a <u>catalyst material</u> is a mixture of substances that includes binder, catalyst, and perfluorovinylether sulfonic acid as disclosed and claimed. The Examiner also disagrees with applicant's interpretation of paragraph 69 on page 18 because this interpretation does not appear to make any sense since it is unclear how the insoluble substance

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would be introduced into the <u>catalyst material</u> and selectively be leached to create voids in the <u>catalyst</u>. As admitted by applicant on page 9 of the remarks filed on 9/12/2005, the <u>catalyst</u> <u>material</u> is a mixture of the <u>catalyst</u> and other materials.

Applicant also asserts without providing any evidence on page 9 that specific methods of putting the insoluble substance into the <u>catalyst material</u> and that methods for leaching the metal zinc, aluminum, or tin out of a <u>material</u>, and method of removing non-ionic surfactant by heating at an elevated temperature are known in the art. Applicant asserts that it is well established that a patent need not teach and preferably omits what is well known in the art on page 9 of the remarks. It appears that applicant is admitting on page 9 of the arguments the putting an insoluble substance into the <u>catalyst material</u> and subsequent removable of the insoluble substance from the <u>catalyst material</u> is well known in the art.

However, applicant's assertion is moot because the comparison in the claims is between the surface area of a <u>catalyst</u> with increased surface area by removing the insoluble substance introduced therein and the surface area of a <u>catalyst</u> not having any insoluble substance introduced therein, not between the surface area of a <u>catalyst material</u> which is a mixture of materials that has been treated with the insoluble substance and the surface area of a <u>catalyst material</u> that has not been treated with the insoluble substance. Nevertheless, for reasons stated in the previous Office Action, the original disclosure is not enabled for temporarily introducing an insoluble substance in a <u>catalyst</u> that is already formed in order to compare its increased surface area with a formed <u>catalyst</u> having no insoluble substance temporarily introduced.

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As mentioned above, applicant states on page 9 of the remarks that the mixing of zinc, aluminum, and tin in other materials and leaching them out are known in the art. It is unclear to the Examiner how mixing in zinc, aluminum, and tin into a <u>catalyst material</u> comprising a <u>catalyst</u>, a binder, and perfluorovinylether sulfonic acid and subsequently leaching of zinc, aluminum, and tin would create voids in the catalyst material since mixing of the metals (zinc, tin, aluminum) into the <u>catalyst material</u> only creates a physical association of the metals with the <u>catalyst</u> and not a chemical one. Since the insoluble substances are not introduced into the <u>catalyst</u> in the <u>catalyst material</u> by mixing, they would not be able to create voids in the <u>catalyst</u> itself after the step of removing the insoluble substance contrary to applicant's assertions on page 10 of the remarks filed on 9/12/2005.

Conclusion

Any inquiry concerning this communication or earlier communications should be directed to examiner Susy Tsang-Foster whose telephone number is (571) 272-1293. The examiner can normally be reached on Monday through Friday from 9:30 AM to 6:00 PM.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Patrick Ryan can be reached at (571) 272-1292.

The fax phone number for the organization where this application or proceeding is assigned is (571) 273-8300.

Information regarding the status of an application may be obtained from the Patent

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/st

SUSYTSANG-FOSTER PRIMARY EXAMINER

Anoy Isang Toster